

UGC NET, GATE, CSIR NET, IIT-JAM, IBPS, CSAT/IAS, CLAT, ISEET, SLET, CTET, TIFR, NIMCET, JEST etc.



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Q 1-25 (1 MARK EACH)

- 1. The product of Reynolds number and Prandtl number is called
 - (A) Rayleigh number
 - (B) Peclet number
 - (C) Stanton number
 - (D) Graetz number
- 2. The power required to crush 1000 tons/h of limestone (work index for limestone = 12.74) if 80 percent of the feed passes a 50.8 mm screen and 80 percent of the product passes a 3.175 mm screen, is
- **3.** The inverse Laplace transform of $\frac{1}{2s^2+3s+1}$ is

(A)
$$e^{-t/2} - e^{-t}$$

- (B) $2e^{-t/2} e^{-1}$
- (C) $c^{-1} 2e^{-t/2}$
- (D) $e^{-t} e^{-t/2}$
- 4. In a countercurrent gas absorber, both the operating and equilibrium relations are linear. The inlet liquid composition and the exit gas composition are maintained constant. In order to increase the absorption factor
 - (A) The liquid flow rate should decrease
 - (B) The gas flow rate should increase
 - (C) The slope of the equilibrium line should increase
 - (D) The slope of the equilibrium line should decrease
- 5. A mixture of nitrogen and acetone vapour at 760 mm Hg and 27°C has a percentage solution of 75%. Its molal absolute humidity is _____.(Vapour pressure of acetone is given by

$$\ell n \ p^{sat} = 14.717 - \frac{297595}{T - 39.52} \big)$$

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exchanger at 100°C and leaves at 60°C. The cold fluid enters the heat exchanger at 40°C. The mean temperature difference between the two fluids is

(A) (100 + 60 + 40) / 3°C

- (B) 60°C
- (C) 45°C
- (D) 20°C
- **11.** Species A is diffusing at steady state from the surface of a sphere (radius = 1 cm) into a statement fluid. If the diffusive flux at a distance r = 3 cm from the centre of the sphere is 27 mol/cm², the diffusive flux in mol/cm² at a distance r = 9 cm is
- **12.** The volume rate of laminar flow of a Newtonian fluid of constant density in a circular tube of diameter d at steady state is proportional to
 - (A) d^2
 - (B) d⁴
 - (C) d³
 - (D) d
- **13.** A well-insulated hemispherical furnace (radius = 1 cm) is shown in Fig. The self view factor of radiation for the curved surface 2 is _____.



- A thin flat plate 2 m × 2 m is hanging freely in air. The temperature of the surroundings is 25°C. Solar radiation is falling on one side of three plate at the rate of 500 W/m². Temperature of the plate will remain constant at 30°C, if the convective heat transfer coefficient (in W/m² °C) is
 - (A) 25

(B) 50

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- 3 (C)
- (D) 1 ε
- 18. The left face of a one dimensional slab of thickness 0.2 m is maintained at 80° C and the right face is exposed to air 30° C. The thermal conductivity of the slab is 1.2 W/(m K) and the heat transfer coefficient from the right face is 10 W / (m² K). At steady state the temperature of the right face in °C is
- 19. Knudsen diffusion coefficient
 - (A) depends on temperature and it is independent of pressure
 - (B) depends on pressure and it is independent of temperature
 - (C) depends on temperature and pressure
 - (D) Is independent of temperature and pressure
- **20.** For the air-water system under adiabatic conditions, the adiabatic saturation temperature and the wet bulb temperature are nearly equal, because
 - (A) water has a high latent heat of evaporation
 - (B) Lewis number is close to unity
 - (C) they are always equal under all circumstances
 - (D) Solubility of the components of air in water is very small.
- 21. Given the following data :
 - Inside heat transfer coefficient = $25 \text{ W} / \text{m}^2 \text{ °C}$
 - Outside heat transfer coefficient = $25 \text{ W} / \text{m}^2 \circ \text{C}$,
 - Thermal conductivity of bricks (15 cm think) = 0.15 W / mk
 - The overall heat transfer coefficient (in W / m² K) will be closer to the
 - (A) inverse of heat transfer coefficient
 - (B) heat transfer coefficient thermal conductivity of bricks

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[For water, the kinematic viscosity, $v = 0.858 \times 10^{-6} \text{ m}^2 / \text{ s.}$].

- (A) 1 m
- (B) 0.43 m
- (C) 43 m
- (D) 103 m

Q 26-55 (2 MARKS EACH)

26. For a pure substance, the Maxwell's relation obtained from the fundamental property relation dU = TdS - PdV is

- (A) $(\partial T/\partial V)_{s} = -(\partial P/\partial S)_{v}$
- (B) $(\partial P/\partial T)_{V} = (\partial S/\partial V)_{T}$
- (C) $(PT/\partial V)_{s} = (\partial V/\partial S)_{P}$
- (D) $(\partial V / \partial T)_{P} = -(\partial S / \partial P)_{T}$
- **27.** In constant pressure filtration, the rate of filtration follows the relation (υ: filtrate volume, t: time and k & c : constants)

28. It takes 6 hours to dry a wet solid from 50% moisture content to the critical mixture content of 15%. Then _____ time it will take to dry the solid to 10% moisture content, under the same

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- **32.** A piece of wood having weight 5 kg floats in water with 60% of its volume under the liquid. The specific gravity of wood is
 - (A) 0.83
 - (B) 0.6
 - (C) 0.4
 - (D) 0.9

33. If the equilibrium curve is concave upward, then the minimum liquid gas ratio for absorption

- (A) Is zero
- (B) Corresponds to an exit liquid concentration in equilibrium with the incoming gas
- (C) Corresponds to an exit liquid concentration in equilibrium with the leaving gas
- (D) Corresponds to an entering liquid concentration in equilibrium with the entering gas
- 34. In liquid-liquid extraction 10 kg of a solution containing 2 kg of solute C and 8 kg of solvent A is brought into contact with 10 kg of solvent B. Solvents A and B are completely immiscible in each other whereas solute C is soluble in both the solvents. The extraction process attains equilibrium. The equilibrium relationship between the two phases is Y* = 0.9X where Y* is kg of C/kg of B and X is kg of C/kg of A. Choose the correct answer
 - (A) The entire amount of C is transferred to solvent B
 - (B) Less then 2 kg but more than 1 kg of C is transferred to solvent B
 - (C) Less then 1 kg of C is transferred to B
 - (D) no amount of C is transferred to B

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- (A) 0.5
- (B) 0.4
- (C) 0.515
- (D) 0.5

36. A ship whose hull length is 100 m is to travel at 10 m/sec. For dynamic similarly, at what velocity should a1 : 25 model be towned through water ?

- (A) 10 m/sec
- (B) 25 m/sec
- (C) 2 m/sec
- (D) 50 m/sec
- **37.** Water flows up a tapered pipe as shown in the figure. What is the magnitude of the deflection h of the differential mercury manometer corresponding to a discharge of 126 L/s ? Friction in the pipe can be completely neglected.

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- (A) 16.28 cm
- (B) 17.28 cm
- (C) 19.28 cm
- (D) 25.28 cm

38. For a flow with $\phi = 3xy$, corresponding value of ϕ will be

- (A) 4 xy + c
- (B) 4 xy + c
- (C) 8 xy + c
- (D) 8 xy + c

39. Extractive distillation is used in the manufacture of

- (A) nitric acid
- (B) caustic soda
- (C) sulphuric acid
- (D) urea
- **40.** For the flow $\phi = x + y + 3$, corresponding value of ϕ will be
 - (A) x + y + c
 - (B) x y + c
 - (C) x y + c
 - (D) x + y + c
- 41. A cylinder of 0.122 m radius rotates concentrically inside a fixed cylinder of 0.128 m radius. Both cylinders are 0.305 m long. What is the viscosity of the liquid that fills the space between the cylinders if a torque of 0.881 N-m is required to maintain an angular velocity of
 - 60 revolutions per minute ?
 - (A) 0.21 Pas
 - (B) 0.23 Pas
 - (C) 0.29 Pas
 - (D) 0.31 Pas

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42. For a guage pressure of A of – 10.89 kPa, what is the specific gravity of the gauge liquid B in the figure below ?

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- (B) 4.573 kPa
- (C) 6.573 kPa
- (D) 7.573 kPa
- **44.** A solid cone of diameter 20 cm and height 16 cm floats with its vertex downwards in water as shown in the figure. This cone is in

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Common Data Q. 50-51

A circular plate of diameter of 0.65 m is immersed in a liquid of relative density 0.85 with its plane making an angle of 30° with the horizontal. Centre of the plate is at a depth of 1.50 m below the free surface.

1.5 m

Y₁

3

- (A) 4142 N
- (B) 6321 N
- (C) 8282 N
- (D) 9296 N
- 51. What is the location of centre of prossure ?
 - (A) 1.0406 m
 - (B) 2.0406 m
 - (C) 3.0406 m
 - (D) 4.0406 m Linked Answers Q. 52 – 53

At 318 K and 24.4 kPa, composition of the system ethanol (1) and toluene (2) at equilibrium is $x_1 = 0.3$ and $y_1 = 0.634$. The saturation pressure at the given temperature for the pure components are $P_1^{S} = 23.06$ kPa and $P_2^{S} = 10.05$ kPa respectively.

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- 52. The liquid- phase activity coefficients will be
 - (A) 0.4084
 - (B) 1.4084
 - (C) 2.2361
 - (D) 1.2964
- **53.** The value of G^E/RT for the liquid phase will be
 - (A) 0.4084
 - (B) 1.4084
 - (C) 2.2964
 - (D) 2.2361

Linked Answer Q. 54-55

Figure below shows a nozzle at the end of a pipe discharging oil from a tank to atmosphere.

The head H in the tank is 4.0 m. The loss in the pipe can be taken as $20 \text{ V}^2/2\text{g}$, where V = velocity in the pipe. The loss of energy in the nozzle can be assumed to be zero.

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- 55. What is the pressure at the base of the nozzle ?
 - (A) 21.2 kPa
 - (B) 31.2 kPa
 - (C) 41.2 kPa
 - (D) 51.2 kPa

GENERAL APTITUDE

Q 56-60 (1 MARK EACH)

- 56. REASON : SFBTPO :: THINK : ?
 - (A) SGHMJ
 - (B) UIJOL
 - (C) UHNKI
 - (D) UJKPM
- 57. MORTAL opposite word -
 - (A) Divine
 - (B) Immortal
 - (C) Spiritual
 - (D) Eternal
- **58.** ALERT similar word
 - (A) Energetic
 - (B) Observant
 - (C) Intelligent
 - (D) Watchful
- **59.** A shopkeeper expects a gain of 22.5% on his cost price. If in a week, his sale was of Rs. 392, what was his profit?
 - (A) Rs. 18.20
 - (B) Rs. 70
 - (C) Rs. 72

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(D) Rs. 88.25

60. IF '+' stands for '-' , '-' stands for 'x', 'x' stands for '÷'and '÷'stands for '+'then what is the value of 56x7÷13-11+15-8÷2-7?

- (A) 30
- (B) 45
- (C) 60
- (D) 90

Q 61-65 (2 MARKS EACH)

61. 'Captain' is related to 'Soldier' in the same way as 'Leader' is related to

- (A) Chair
- (B) Followers
- (C) Party
- (D) Minister
- 62. PORK:PIG
 - (A) rooster:chicken
 - (B) mutton:sheep
 - (C) steer:beef
 - (D) lobster:crustacean
- 63. My uncle decided to take and my sister to the market.
 - (A) I

(B) mine

(C) me

- (D) myself
- 64. Answer the question based on the given line graph.

Ratio of Exports to Imports (in terms of money in Rs. crores) of Two Companies Over the Years

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Answer key

Question	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20
Answer	в	1696 Kw	A	D	295.3 K or 22.3°C	A	D	С	<1	D	3	В	1/2	A	2.62 and 3.68 min ⁻¹	c	►B	48.7 °C	A	в
Question	21	22	23	24	25	26	27	28	29	30	31	32	33	34	35	36	37	38	39	40
Answer	D	Α	D	-2,-3	В	Α	В	1.1882	В	D	С	В	D	В	С	С	С	D	А	В
Question	41	42	43	44	45	46	47	48	49	50	51	52	53	54	.55	56	57	58	59	60
Answer	В	Α	Α	Α	В	В	В	В	В	Α	С	Α	A	С	В	В	В	D	С	В
Question	61	62	63	64	65															
Answer	В	В	С	В	В															
HINTS AND SOLUTIONS 1.(B) Peclet number is the product of Reynolds number and prandtl number $N_{Pe} = N_{R_2} \times N_{P_r}$ where																				
2 . 1	(Power required) $\left(\frac{P}{m}\right) = 0.3162 W_i \left(\frac{1}{\sqrt{D_{Pb}}} - \frac{1}{\sqrt{D_{Pa}}}\right)$																			

 $= 1000 \times 0.3162 \times 12.74 \left(\frac{1}{\sqrt{3.175}} - \frac{1}{\sqrt{50.8}}\right)$

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= 1696 kW

1

 $\frac{1}{2s^2+3s+1}$

Given Laplace =

3.(A)

PM ULASSE

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$$= L^{-1} \left\{ \frac{1}{2s^2 + 3s + 1} \right\}$$
$$= L^{-1} \left\{ \frac{1}{2s^2 + 25 + s + 1} \right\}$$
$$= L^{-1} \frac{1}{(2s + 1)(s + 1)}$$

on partial fraction

$$= L^{-1} \left\{ -\frac{1}{(s+1)} + \frac{1}{2} \left(\frac{1}{25+1} \right) \right\}$$

On separating

$$= L^{-1} \left\{ \frac{-1}{(s+1)} \right\} + \frac{1}{2} L^{-1} \left\{ \frac{1}{2s+1} \right\}$$
$$= -e^{-t} + \frac{1}{2} \frac{e^{-\frac{1}{2}}}{\frac{1}{2}}$$
$$= -e^{-t} + e^{-t/2}$$
$$= e^{-t/2} - e^{-t}$$

4.(D) Absorption factor $\left(\frac{L}{mG}\right)$ is the ratio of the slope of operating line to the equilibrium curve.

So, to increase Absorption factor , Slope of equilibrium curve decreases.

Percentage saturation = 75%

- P_A^{sat} = Vapour pressure of acetone at 27°C = 33.358 kPa
- (a) Saturated molal absolute humidity,

 $Y_{s} = \frac{P_{A}^{sat}}{P_{t} - P_{A}^{sat}} = \frac{33.358}{101.325 - 33.358} = 0.4908 \frac{\text{kmol acetone vapour}}{\text{kmol nitrogen}}$ Ans. Percentage saturation $=\frac{Y}{Y} \times 100 = 75$ $Y = \frac{75}{100} Y_s = 0.75 \times 0.4908$ or, kmol acetone vapour = 0.3681 kmol nitrogen (b) Absolute humidity , $Y' = Y \frac{M_A}{M_B} = 0.3681 \times \frac{58}{28} = 0.7625 \frac{\text{kg} \text{ acetone}}{\text{kg nitrogen}}$ $Y = \frac{P_A}{P_A - P_A}$ (C) $p_{A} = \frac{YP_{t}}{Y+1} = \frac{0.3681 \times 101.325}{0.3681 + 1} = 27.262 \text{ kPa}$ *:*. $= \frac{P_A}{P_A^{\text{sat}}} \times 100 = \frac{27.262}{33.358} \times 100 = 81.72\%$ (d) Relative humidity (e) Volume percent acetone = $\frac{P_A}{P_t} \times 100 = \frac{27.262}{1.01.325} \times 100 = 26.90\%$ (f) At dew point, $p_A = P_A^{sa}$ $\ln (27.262) = 14.7171 - \frac{2975.95}{T - 34.5228}$ *.*.. T = dew point = 295.3 K or 22.3°C Moisture content on wet basis = $\left(\frac{\text{kg moisture}}{\text{kg dry solid} + \text{kg moisture}}\right) \times 100$ 6.(A) $=\frac{100x}{1+x}$ Moisture content on Dry basis = $\frac{ky \text{ moisture}}{kg \text{ dry solid}} \times 100$

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$$4\pi r^{2} N_{Ar} | r = 4\pi r^{2} N_{A}|_{rxc}$$
or
$$\frac{d}{dt} (4\pi r^{2} N_{Ar}) = 0 \Rightarrow 4\pi r^{2} N_{Ar} = C \text{ (constant)}$$

$$4\pi r^{2} N_{Ar} = 4\pi r^{2} N_{Ar2} \Rightarrow \frac{N_{Ar2}}{N_{Ar1}} = \left(\frac{r}{r_{2}}\right)^{2}$$

$$\frac{N_{Ar2}}{2T} = \left(\frac{3}{3}\right)^{2} \Rightarrow N_{Ar/2} = 3 \text{ mol / cm}^{2}$$
12.(B) By Hagen - poiseulle equation for laminar flow in circular pipe.
$$AP = \frac{32\mu VL}{D^{2}} = \frac{128\mu QL}{D^{4}}$$

$$\frac{AP}{L} = \frac{128\mu QL}{D^{4}}$$

$$\frac{AP}{L^{2}} = \frac{128\mu QL}{D^{4}}$$

$$\frac{Q \alpha D^{4}}{D^{4}}$$
13. 1/2
$$F_{11} + F_{12} = 1$$

$$F_{12} = 1 - F_{11}$$

$$= 1 - 9$$

$$= 1$$

$$Ar F = \frac{2}{2} F_{21}$$

$$F_{11} + F_{12} = 1$$

$$F_{12} = 1 - F_{11}$$

$$= 1 - 9$$

$$F_{11} + F_{12} = 1$$

$$F_{12} = 1 - F_{11}$$

$$= 1 - 9$$

$$F_{11} + F_{12} = 1$$

$$F_{12} = 1 - F_{11}$$

$$= 1 - 9$$

$$F_{11} + F_{12} = 1$$

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$$C_A = 1 + \frac{F}{V}$$

component A mass balance:

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18. 48.7

Let the temperature of the right face at steady state be T°C.

$$q = -1.2 \quad \frac{(T-80)}{0.2} = 10(T-30)$$

or,

19.(A) Diffusion in porous solid is given

 $D_{\kappa,A}$ knudsen diffusion coefficient

$$D_{K_1A} \alpha \left(\frac{T}{M_A}\right)^{1/2}$$

So, it is dependent on temperature but independent of pressure.

- **20.(B)** When adiabatic saturation temperature and wet bulb temperature are equal. Both thermal diffusivity and mass transfer diffusivity is equal
 - i.e.

Le =
$$\frac{L}{D}$$
 = 1

21.(D) Overall heat transfer coefficient **U** is given by the relation

 $\frac{1}{U} = \frac{1}{h_a} + \frac{t}{k} \frac{1}{h_b}$

where

 h_a and h_b = inside and outside heat transfer coefficients (convective),

- thickness of bricks

coefficient of thermal conductivity of the bricks alone.

22.(A)

General heat conduction equation is,

k

+ 3x + 16

$$\frac{1}{\alpha}\frac{\partial T}{\partial \tau} = \left(\frac{\partial^2 t}{\partial x^2} + \frac{\partial^2 t}{\partial y^2} + \frac{\partial t}{\partial x^2}\right) + \frac{Q}{k}$$

Without heat generation (Q = 0) and in x-direction only,

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$$\begin{aligned} \frac{1}{\alpha} \frac{\partial T}{\partial \tau} &= \frac{\partial^2 t}{\partial x^2} \\ \text{or} \qquad & \frac{\partial T}{\partial \tau} = \alpha \cdot \frac{\partial^2 t}{\partial x^2} \\ &= (0.0003) \ (6) = -0.00018 \ \text{K/s} \end{aligned}$$
23.(D) d is the correct option as
Naphtha is made from ethylene
Methanol is used for producing synthesis gas.

$$CH_3 \ OH \xrightarrow{\longrightarrow} CO + 2H_2$$
Naphthalene is prepared from pthalic anhydride
24. $-2, -3$
Given transfer function $K \frac{(S+1)(S+4)}{(S+2)(S+3)}$
for poles of transfer function $S = -2$ $S = -3$
25.(B) $R_{ex} = \frac{xU}{v}$
 $\therefore x = \frac{5 \times 10^5 \times 00858 \times 10^{-6}}{1} = 0.4293 \approx 0.43$
26.(A) Maxwell relation
 $m_1 = TdS \le PdV$
It follow exact differential equation
 $dF = Mdx + Ndy$
 $\left(\frac{\partial M}{\partial y}\right)_{s} = \left(\frac{\partial N}{\partial x}\right)_{s}$

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$$Q = \frac{A\sigma(T_1^4 - T_1^4)}{\left(\frac{1}{\epsilon_1} - 1\right) + 1 + \left(\frac{1}{\epsilon_2} - 1\right)}$$

$$= \frac{5.67 \times 10^{-6} (800^4 - 500^4)}{\left(\frac{1}{0.8} - 1\right) + 1 + \left(\frac{1}{0.6} - 1\right)} = 10.313 \text{ kW/m}^2$$

30.(D) Mean temperature of cold fluid

$$=\frac{30+50}{2}=40^{\circ} \text{ C}$$

Mean temperature of hold fluid

$$= \frac{130 + 150}{2} = 140^{\circ} \text{ C}$$

Mean temperature difference

$$= 140 - 40 = 100^{\circ}$$

31.(C) Pressure exerted on a certain area is equal to the force applied.

So,
$$(\Delta p) \left(\frac{\pi}{4}D^{2}\right) = F$$

 $\Rightarrow \left(\frac{4fLP\overline{V}^{2}}{2D}\right) \left(\frac{\pi}{4}D^{2}\right) = F$
 $\Rightarrow \left(\frac{4LP\overline{V}^{2}}{2D}\right) \left(\frac{\pi}{4}D^{2}\right) = F$
 $\Rightarrow \left(\frac{4LP\overline{V}^{2}}{2D}\right) \left(\frac{\pi}{4}D^{2}\right) = F$
 $\Rightarrow \left(\frac{4LP\overline{V}^{2}}{2D}\right) \left(\frac{\pi}{4}D^{2}\right) = F$
 $\Rightarrow F$
 $= F$
 $= \left[\frac{8\pi\mu L^{2}}{2}\right]_{L/2}^{L} = Ft$
 $\Rightarrow T = \frac{6\pi\mu L^{2}}{F}$

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or
$$y^3 = H^3S$$

 \therefore $y = H(S)^{1/3} = 16 \times (0.8)^{1/3} = 14.853 \text{ cm}$
If B is the centre of buoyancy, then
 $\Theta = \frac{3}{4} y = 0.75 \times 14.853 = 11.1398 \text{ cm}$
 $\Theta = \frac{3}{4} H = 0.75 \times 16 = 12 \text{ cm}$
Diameter at water surface,
 $d = 2y \tan \theta = 18.566 \text{ cm}$
 $BM = \frac{1}{V} = \frac{\pi d^4/64}{\frac{1}{3}(\frac{\pi d^4}{4})y} = \frac{3}{16}(\frac{y^2}{4}) = \frac{3}{4}y$
 $\tan^2 \theta \cdot 41.351 \text{ m}$
 $OM = B + MB + 11.1398 \oplus 4.351 \text{ f}$ 15.491 cm
 $\Theta = 12.00$
 \therefore $MG = (15.491 + 12.00) = 3.491 \text{ cm}$
(i.e. M is above G byta 491 cm)
Hence the cone is under stable equilibrium.
45.(5) Velopity grediem $= \frac{dH}{dy} = 0.25 \text{ m/sec meter}$
Dansity, $p = 129.3 \text{ slug/meter}^3$
Kinematic viscosity, $v = 6.30 \times 10^{-4} \text{ meter}^2/\text{sec}$
 $Shear stress = \mu \frac{dU}{dy} = \rho v \frac{dU}{dy}$
 $= 129.3 \times 6.30 \times 10^{-4} \times 0.25$
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UGC NET, GATE, CSIR NET, IIT-JAM, IBPS, CSAT/IAS, CLAT, ISEET, SLET, CTET, TIFR, NIMCET, JEST etc. $x_2 = \frac{0.14 \times 0.5}{0.7} = 0.1 \text{ m}$:. **48.(B)** u = 3x and u = -3yEquation of a streamline in two-dimensional flow $\frac{dx}{dt} = \frac{dy}{dt}$ $\frac{dx}{3x} = -\frac{dy}{3y}$ Here Integrating, $\frac{1}{3} \ln x = -\frac{1}{3} \ln y + \frac{1}{3} \ln c$ where c = constant In xy = In c :. or xν = C For the streamline passing through (1, 1), c = (1, 1)Hence required streamline equation is, **49.(B)** $u = -y^3$ and v = -6xDifferential equation of the streamline is, dy Integrating, ∫6x dx = ∫y² dy + c $3x^2 - \frac{y^3}{2} = c$ or x = 1, y = 1, we getPutting c = $3 - \frac{1}{3} = 2\frac{2}{3}$

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$$y_i^{P} = \gamma_i x_i^{P} P_i^{S}$$

 $\gamma_1 = \frac{y_1^{P}}{x_1^{P}_1^{S}} = \frac{0.634 \times 24.4}{0.3 \times 23.06} = 2.2361,$

$$\gamma_2 = \frac{y_2 P}{x_2 P_2^s} = \frac{0.366 \times 24.4}{0.7 \times 10.05} = 1.2694$$

53.(A) Excess free energy is related to the activity coefficient and the composition by

$$\frac{C^{E}}{RT} = x_1 \ln \gamma_1 + x_2 \ln \gamma_2$$

54.(C) By continuity equation,

Q = V₂
$$\frac{\pi}{2}$$
 (D)² = V₃ $\frac{\pi}{2}$ (d)²

Velocity of the pipe,

$$V_2 = V_3 \left(\frac{d}{D}\right)^2 = \left(\frac{25}{200}\right)^2 V_3 = \frac{1}{64} V_3$$

Applying Bernoulli's equation to points 1 and 3 with the centre line of the pipe as datum and atmospheric pressure as zero. The velocity at point 1 can be taken as zero.

$$p_{1} + V_{1}^{2} + z = p_{3} + \frac{V_{3}^{2}}{2g} + z_{3} + H_{L}$$
of
$$0 + 0 + H = 0 + \frac{V_{3}^{2}}{2g} + 0 + 20 \frac{V_{2}^{2}}{2g}$$

$$H = \frac{V_{3}^{2}}{2g} + 20 \times \left(\frac{1}{64}\right)^{2} \frac{V_{3}^{2}}{2g} = 1.004 \frac{V_{3}^{2}}{2g}$$

But H = 4.0 m, therefore

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$$V_3 = \left(\frac{2 \times 9.81 \times 4.0}{1.004}\right)^{1/2} = 8.83 \text{ m/s}$$

Discharge, Q = $\frac{\pi}{4} \times (0.025)^2 \times 8.83$

$$= 4.33 \times 10^{-3} \text{ m}^{3}/\text{s}$$

55.(B) Velocity in the pipe,
$$V_2 = \frac{1}{64} \times 8.83 = 0.13$$
 m/s

Loss of head in the pipe,

$$H_L = 20 \times \frac{V_2^2}{2g} = 20 \times \frac{(0.13)^2}{2 \times 9.81} = 0.019m$$

Applying Bernoulli's equation to points 1 and

$$0 + 0 + H = \frac{p_2}{\gamma} + \frac{V_2^2}{2g} + 0 + H_L$$

$$4.0 = \frac{p_2}{\gamma} = \frac{(0.137)^2}{2 \times 9.81} + 0 + 0.019$$

or

or

or

56.(B)

$$\gamma = \frac{0.8 \times 998 \times 9.81}{1000} = 7.832 \text{ kN/m}^3$$

4.0 – 0.0199 = 3.980 m

... Pressure at the base of the nozzle,

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As	Similarly
R <mark>+1</mark> → S	T +1 ∪
E +1 → F	H +1 → I
A +1 → B	I J
s _+1 →⊤	N +1 → 0
0 _+1 ₽	K _+1 → L
N +1 → O	

- **57.(B)** Mortal means causing or capable of causing death while Immortal means one who is not subject to death.
- 58.(D) Alert means engaged in or accustomed to close observation, ie. Watchfulness.

59.(C) C.P. = Rs.
$$\left(\frac{100}{122.5} \times 392\right) = \text{Rs}\left(\frac{1000}{1225} \times 392\right) = 320\text{Rs}$$

Profit = Rs. (392 - 320) = Rs. 72.

60.(B) Changing the symbols as given in the problem the above expression is 56÷7+13x11-15x8+2x7

Solving the BODMAS rule, we get 8+143-120+14=165-120=45

- **61.(B)** 'Captain' is supposed to lead the battalian of 'Soldiers' int he same way as 'Leader' is supposed to lead the 'Followers'
- **62.(B)** Pork is meat from a domestic hog or pig. Similarly, mutton is meat from a mature domestic sheep.
- 63.(C) My uncle decided to take me and my sister to the market.
- **64.(B)** The exports are more than imports in those years for which the exports to imports ratio are more than 1. For Company A, such years are 1995, 1996 and 1997.

Thus, during these 3 years, the exports are more than the imports for Company A.

65.(B) This is a simple subtraction series. Each number is 6 less than the previous number.

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